

**Metropolitan Museum of Art**  
**Gas Chromatography- Mass Spectrometry (GC-MS) Results from Material Analysis**

This document includes (1) a mass spectrum and (2) the volatile organic compounds (VOCs) emitted from samples using GC-MS analysis. The data is not interpreted; however, several classes of chemicals are highlighted because they are potential risks for artwork in an enclosed environment. A basic key, provided below, indicates those classes. The amount of each chemical identified has not been determined; similarly, it is not known how much of each chemical is necessary to do damage to art. Finally, peaks may be present that are the result of the sample adsorbing chemicals from the air and reemitting them during testing rather than being inherent to the sample. Research is ongoing to determine specifically which chemicals and amounts are required to negatively affect artifacts.

**Highlighted data:**

Pink – chemicals currently known to be hazardous to art

Green – amines; can raise the pH, are suspected to react with acids and may form crystals in an enclosed environment

Yellow – chemicals of the following type, which *may* be hazardous to art:

*Acids* – lower the pH, corrosive to metals, degrade organic materials

*Aldehydes* – can convert to acids with heat or exposure to UV light

*Esters* – can hydrolyze into acids with heat and humidity

*Sulfur-containing compounds* – known to tarnish and corrode some metals

*Halogenated compounds* – can become reactive with exposure to heat and UV light

*Nitrogen-containing, not amine* – can react with other off-gassed chemicals

*Alkynes* – can become reactive when exposed to heat or UV light

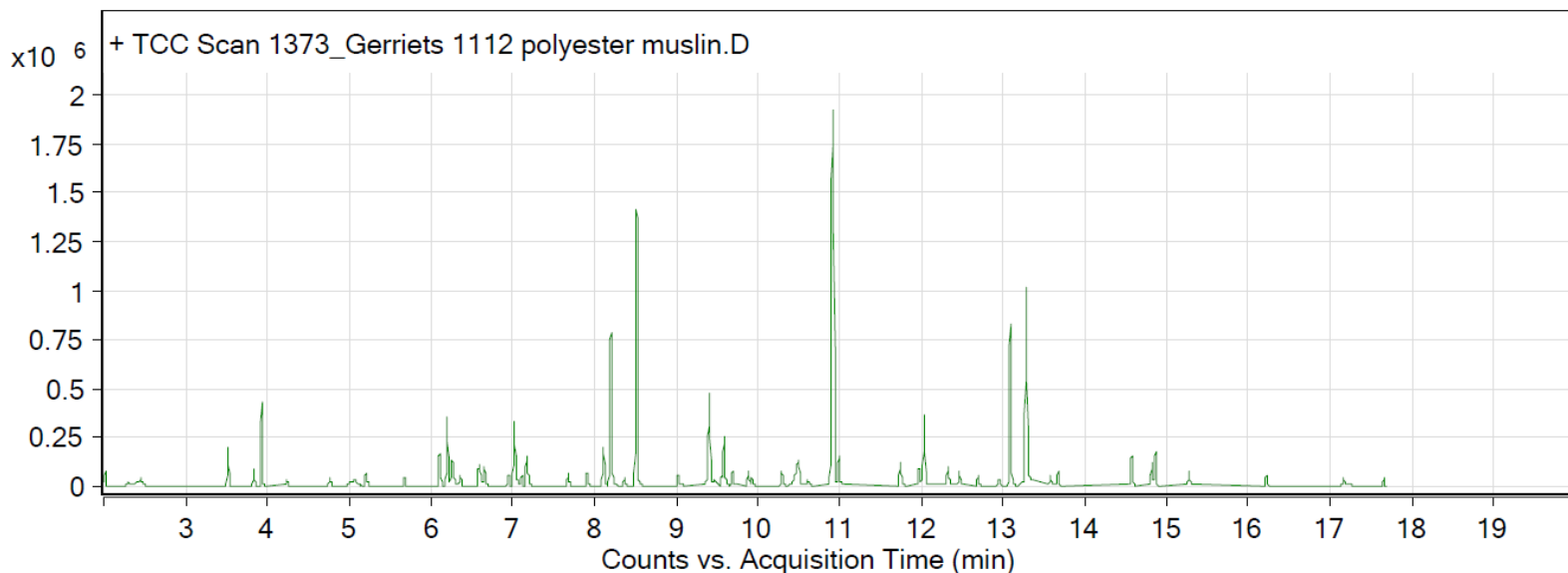
Sample: Gerriets 1112 IFR Polyester muslin CS

Oddy test score: Unsuitable

Date collected: 08/15/2016

Technique used: SPME with a PDMS/DVB fiber; Agilent 7890B GC and 5977B MS fitted with a GL Sciences OPTIC-4 multimode inlet and LEAP PAL RTC autosampler; Pre-heated at 60°C for 20 minutes; fiber exposure at 60°C for 20 minutes; sample injected into 220°C inlet and cotrapped for 2 min at -15°C; GC ramped from 40°C to 225 °C at 10°C/min. Data analyzed in masshunter Qualitative. Samples > 90% match with a NIST 17.0 library are reported.

VOCs not highlighted are because they were also observed in blanks: (1) 12.0 min: 2-methyl-, 3-hydroxyl-2,4,4-trimethylpentyl ester propanoic acid



2.01	Acetone	C3H6O
3.51	Benzene, methyl-	C7H8
3.83	Hexanal	C6H12O
3.93	Cyclotrisiloxane, hexamethyl-	C6H18O3Si3
4.24	2-Furancarboxaldehyde	C5H4O2
4.76	Benzene, 1,3-dimethyl-	C8H10
5.2	Ethanol, 2-butoxy-	C6H14O2
6.1	Benzaldehyde	C7H6O
6.2	Cyclotetrasiloxane, octamethyl-	C8H24O4Si4
6.26	Phenol	C6H6O
6.58	x - ethyl - x - methyl - benzene	C9H12
6.95	3-Cyclohexene-1-carbonitrile	C7H9N
7.02	1-Hexanol, 2-ethyl-	C8H18O
7.17	Benzenemethanol	C7H8O
8.11	Pentadecane	C15H32
8.2	Nonanal	C9H18O
9.02	2-Nonenal, (E)-	C9H16O
9.41	Ethanol, 2-(2-butoxyethoxy)-	C8H18O3
9.92	Ethanol, 2-phenoxy-	C8H10O2
10.29	1-Phenoxypropan-2-ol	C9H12O2
10.99	Tridecane	C13H28
11.97	1-Undecanol	C11H24O
12.03	Propanoic acid, 2-methyl-, 3-hydroxy-2,4,4-trimethylpentyl ester	C12H24O3
13.28	1-Dodecanol	C12H26O
13.58	Heptadecane, 7-methyl-	C18H38
14.57	PENTAN-1,3-DIOLDIISOBUTYRATE, 2,2,4-TRIMETHYL-	C16H30O4
17.65	1,2-Benzenedicarboxylic acid, dibutyl ester	C16H22O4