Metropolitan Museum of Art Gas Chromatography- Mass Spectrometry (GC-MS) Results from Material Analysis

This document includes (1) a mass spectrum and (2) the volatile organic compounds (VOCs) emitted from samples using GC-MS analysis. The data is not interpreted; however, several classes of chemicals are highlighted because they are potential risks for artwork in an enclosed environment. A basic key, provided below, indicates those classes. The amount of each chemical identified has not been determined; similarly, it is not known how much of each chemical is necessary to do damage to art. Finally, peaks may be present that are the result of the sample adsorbing chemicals from the air and reemitting them during testing rather than being inherent to the sample. Research is ongoing to determine specifically which chemicals and amounts are required to negatively affect artifacts.

Highlighted data:

- Pink chemicals currently known to be hazardous to art
- Green amines; can raise the pH, are suspected to react with acids and may form crystals in an enclosed environment

Yellow – chemicals of the following type, which may be hazardous to art:

Acids – lower the pH, corrosive to metals, degrade organic materials

Aldehydes - can convert to acids with heat or exposure to UV light

Esters - can hydrolyze into acids with heat and humidity

Sulfur-containing compounds – known to tarnish and corrode some metals

Halogenated compounds - can become reactive with exposure to heat and UV light

Nitrogen-containing, not amine – can react with other off-gassed chemicals

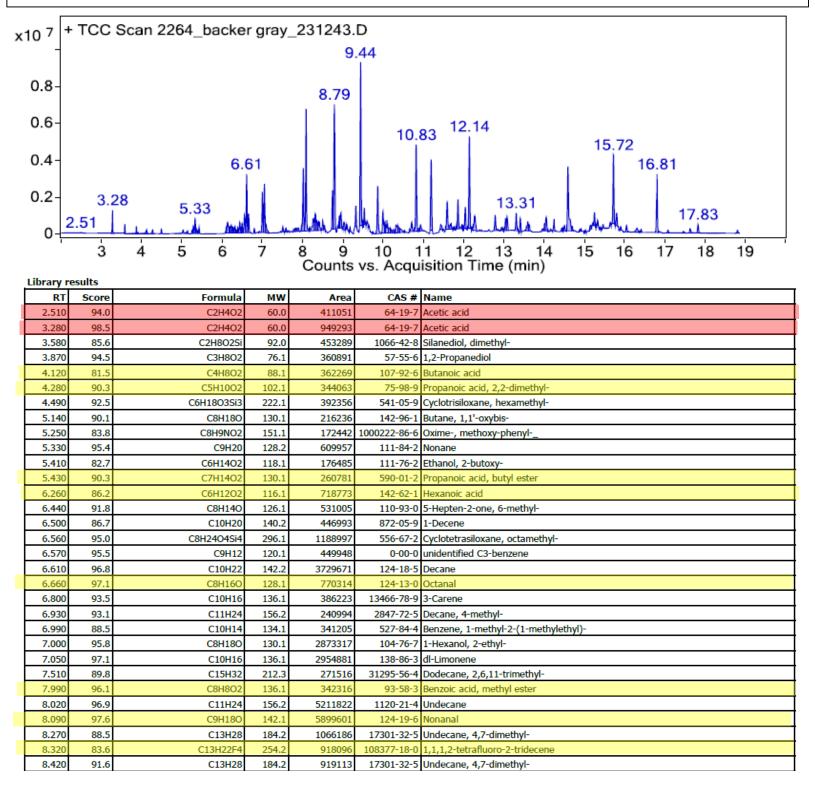
Alkynes - can become reactive when exposed to heat or UV light

Sample: Weatherall Company Inc.; Alcot Plastics backer round rod gray

Date collected: 08/17/2018

Technique used: SPME with a PDMS/DVB fiber; Agilent 7890B GC and 5977B MS fitted with a GL Sciences OPTIC-4 multimode inlet and LEAP PAL RTC autosampler; Pre-heated at 60°C for 20 minutes; fiber exposure at 60°C for 20 minutes; sample injected into 220°C inlet and crotrapped for 2 min at -15°C; GC ramped from 40°C to 225 °C at 10°C/min. Data analyzed in masshunter Qualitative. Samples > 80% match with a NIST 17.0 library are reported.

VOCs not highlighted are because they were also observed in blanks: : (1) 5.3 min: methoxyphenyl oxime; (2) 11.6 min: 2-methyl-, 2,2-dimethyl-1-(2-hydroxyl-1-methylethyl) propyl ester propanoic acid; (3) 11.9 min: 2-methyl-, 3-hydroxyl-2,4,4-trimethylpentyl ester propanoic acid



8.500	90.8	C13H28	184.2	320362	17301-33-6	Undecane, 4,8-dimethyl-
8.740	96.0	C10H20O2	172.1	3000869	103-09-3	Acetic acid, 2-ethylhexyl ester
8.790	94.1	C10H30O5Si5	370.1	9079550		Cyclopentasiloxane, decamethyl-
8.950	81.1	C10H20O2	172.1	1229869		Acetic acid, octyl ester
9.030	81.1	C18H38	254.3	787059		9-methylheptadecane
9.100	86.4	C10H20O	156.2	698270		Cyclohexanol, 5-methyl-2-(1-methylethyl)-, [1S-(1.alpha.,2.alpha.,5.beta.)]-
9.180	86.4	C10H20O2	172.1	387271		Acetic acid, octyl ester
9.290	89.9	C10H8	128.1	291587		1H-Indene, 1-methylene-
9.330	96.0	C12H24	168.2	2344111		1-Dodecene
9.440	95.4	C12H26	170.2	15069366		Dodecane
9.530	96.5	C10H20O	156.2	1835229	112-31-2	
9.560	80.3	C4H8O2S	120.0	263035		Thiophene, tetrahydro-, 1,1-dioxide
9.600	81.2	C12H24O2	200.2	247838		Acetic acid, decyl ester
9.750	92.7	C8H10O2	138.1	201355		Ethanol, 2-phenoxy-
9.870	95.2	C11H20O2	184.1	4100820		2-Ethylhexyl acrylate
10.000	90.3	C11H22O2	186.2	1900371		2-Ethyl-1-hexyl propionate
10.060	83.6	C12H24	168.2	587984		Cyclododecane
10.320	86.5	C18H38O	270.3	804782		Decyl octyl ether
10.430	81.1	C18H38O	270.3	534433		Decyl octyl ether
10.710	94.4	C13H26	182.2	1174130		1-Tridecene
10.820	94.9	C13H28	184.2	7823815		Tridecane
10.930	84.7	C11H22O	170.2	790729		Undecanal
11.200	95.3	C12H36O6Si6	444.1	7511209		Cyclohexasiloxane, dodecamethyl-
11.430	83.7	C13H26	182.2	510029	5617-41-4	Heptylcyclohexane
11.590	91.1	C12H24O3	216.2	1698992		Propanoic acid, 2-methyl-, 2,2-dimethyl-1-(2-hydroxy-1-methylethyl)propyl ester
11.760	88.1	C20H42	282.3	508822		9-methylnonadecane
11.860	92.6	C12H24O3	216.2	3117262		Propanoic acid, 2-methyl-, 3-hydroxy-2,4,4-trimethylpentyl ester
12.040	95.9	C14H28	196.2	2225948		1-Tetradecene
12.150	95.0	C14H30	198.2	8566636		Tetradecane
12.220	81.1	C16H34	226.3	483476		Tridecane, 6-propyl-
12.270	96.8	C12H240	184.2	667772		Dodecanal
12.790	90.5	C14H300	214.2	1581061		1-Tetradecanol
12.950	90.2	C20H42	282.3	524532	112-95-8	
13.310	95.6	C15H30	210.2	1543491		1-Pentadecene
13.600	88.0	C16H34	226.3	406048		Tetradecane, 2,2-dimethyl-
13.630	93.0	C15H240	220.2	285081		Phenol, 2,6-bis(1,1-dimethylethyl)-4-methyl-
13.960	82.4	C13H28	184.2	269283		Nonane, 2-methyl-5-propyl-
14.060	88.5	C15H30 C23H48	210.2	823238		n-Nonylcyclohexane Tricosane
14.160	87.1		324.4	334608		Pentadecane, 3-methyl-
14.250	91.1 87.2	C16H34	226.3 140.2	432061		Pentadecane, 3-methyl- Cyclopentane, 1,2-dimethyl-3-(1-methylethyl)-
14.440 14.510	87.2 92.2	C10H20 C19H38	266.3	316996 532030		Cyclopentane, 1,2-dimethyl-3-(1-methylethyl)- 1-Nonadecene
14.510	92.2 90.4	C19H38 C16H34	200.3	4221927		Hexadecane
14.920	90.4 92.6	C16H34 C15H30O2	242.2	4221927		Dodecanoic acid, 1-methylethyl ester
15.030	92.0	C13H100	182.1	376437		Methanone, diphenyl-
13.030	50.3	C13H100	102.1	370437	119-01-9	
15.250	91.0	C16H34O	242.3	964367	36653-82-4	1-Hexadecanol
15.330	93.3	C16H34O	242.3	626039	629-82-3	Octane, 1,1'-oxybis-
15.410	84.5	C16H48O8Si8	592.2	231193	556-68-3	Cyclooctasiloxane, hexadecamethyl-
15.730	95.5	C17H36	240.3	2716560	629-78-7	Heptadecane
16.050	91.7	C18H20	236.2	497965	3910-35-8	1H-Indene, 2,3-dihydro-1,1,3-trimethyl-3-phenyl-
16.410	88.9	C17H34	238.3	286677	54105-66-7	Cyclohexane, undecyl-
16.800	93.2	C18H38	254.3	3244239	593-45-3	Octadecane
16.860	80.5	C19H40	268.3	199283	1000360-41-0	5,5-Diethylpentadecane
17.080	91.3	C17H34O2	270.3	239210	110-27-0	Isopropyl myristate
	94.4	C16H34O	242.3	421338		1-Hexadecanol
17.630				700000	600 00 F	Nonadecane
17.630 17.830	89.8	C19H40	268.3	783822	029-92-5	Nonadecane