## Metropolitan Museum of Art Gas Chromatography- Mass Spectrometry (GC-MS) Results from Material Analysis

This document includes (1) a mass spectrum and (2) the volatile organic compounds (VOCs) emitted from samples using GC-MS analysis. The data is not interpreted; however, several classes of chemicals are highlighted because they are potential risks for artwork in an enclosed environment. A basic key, provided below, indicates those classes. The amount of each chemical identified has not been determined; similarly, it is not known how much of each chemical is necessary to do damage to art. Finally, peaks may be present that are the result of the sample adsorbing chemicals from the air and reemitting them during testing rather than being inherent to the sample. Research is ongoing to determine specifically which chemicals and amounts are required to negatively affect artifacts.

## Highlighted data:

- Pink chemicals currently known to be hazardous to art
- Green amines; can raise the pH, are suspected to react with acids and may form crystals in an enclosed environment

Yellow – chemicals of the following type, which may be hazardous to art:

Acids – lower the pH, corrosive to metals, degrade organic materials

Aldehydes - can convert to acids with heat or exposure to UV light

Esters - can hydrolyze into acids with heat and humidity

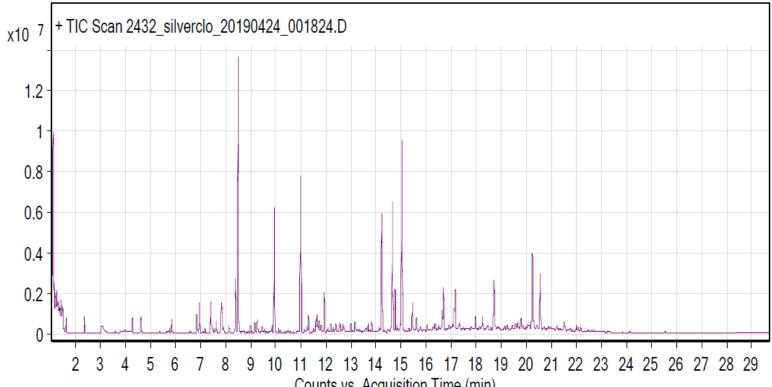
*Sulfur-containing compounds* – known to tarnish and corrode some metals

Halogenated compounds - can become reactive with exposure to heat and UV light

*Nitrogen-containing, not amine* – can react with other off-gassed chemicals

Alkynes - can become reactive when exposed to heat or UV light

Sample: Gaylord Pacific Silvercloth Date collected: 4/24/2019 Oddy test result: Temporary Technique used: SPME with a PDMS/DVB fiber; Agilent 7890B GC and 5977B MS fitted with a GL Sciences OPTIC-4 multimode inlet and LEAP PAL RTC autosampler; Pre-heated at 60°C for 20 minutes; fiber exposure at 60°C for 20 minutes; sample injected into 220°C inlet and crotrapped for 2 min at -15°C; GC ramped from 35°C to 225 °C at 7.5°C/min. Data analyzed in masshunter Qualitative. Samples > 90% match with a NIST 17.0 library are reported. VOCs not highlighted are because they were also observed in blanks: (1) 15.0 min: 2-methyl-, 3-hydroxyl-2,4,4-trimethylpentyl ester propanoic acid



Counts vs. Acquisition Time (min)

Compound Table

## RT Score (Lib) Area Name 3.07 93 2439183 Hexanal 91.66 3.98 625696 2-Furancarboxaldehyde 1333657 2-Pentanone, 4-hydroxy-4-methyl-4.61 95.2 5.84 95.74 1003507 Ethanol, 2-butoxy-6.84 94.66 1546314 Hexanal, 2-ethyl-6.95 2380584 N-benzylidene-dimethylammonium chloride 98.13 7.4 95.67 2326208 Phenol 7.46 90.66 886180 Hexanoic acid 7.62 92.08 815440 Furan, 2-pentyl-7.85 92.95 1806106 Octanal dipropylene glycol monomethyl ether isomer, 97.81 821047 8.13 STRUCTURE UNKNOWN 8.38 95.28 2523387 dl-Limonene 8.39 3132255 1-Hexanol, 2-ethyl-96.15 96.46 26559665 Benzyl alcohol 8.5 9.15 94.39 803078 Ethanone, 1-phenyl-9.25 94.53 1148485 1-Octanol 9.94 11019943 Nonanal 96.77 90.74 13511886 Cyclopentasiloxane, decamethyl-10.99 1407706 Cyclohexanol, 5-methyl-2-(1-methylethyl)-11.3 93.39 11.61 90.39 1543738 Ethanol, 2-(2-butoxyethoxy)-91.67 761314 Benzoic acid, 2-hydroxy-, methyl ester 11.73

11.93	97.01	3487900	Decanal
12.4	90.61	782247	2-Propenoic acid, 2-ethylhexyl ester
13.16	93.89	812837	1-Decanol
13.68	95.89	671532	Tetradecane
15.04	94.63	19013350	Propanoic acid, 2-methyl-, 3-hydroxy-2,4,4- trimethylpentyl ester
15.45	94.39	2511936	Eicosane
15.61	93.5	1229795	Dodecanal
17.12	93.18	1999918	Tridecane